

PY4T01 Condensed Matter Theory

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1. Introduction: What Hamiltonian? Born-Oppenheimer Approximation
 2. Tight-binding method for molecules
 3. From molecules to solid
 4. Going from 1D to 2D and 3D
 5. C-graphite and C-nanotubes
 6. Interfaces: charge transfer and screening
 7. A proper calculation: the Coulomb potential back
 8. The Many-Body problem
 9. The Hubbard model
 10. The mean field approximation
 11. Functional Theory
 12. Density Functional Theory on a Lattice
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1. Electronic transport: Classical model
 2. Quantum Transport
 3. Transfer matrix theory
 4. Disorder and localization

Text Books

General Text:

1. *Electronic Structure of Materials*, A. P. Sutton, (Oxford University Press, 1996, ISBN 0-19-851754-8)
2. *Bonding and Structure of Molecules and Solids*, D. G. Pettifor, (Oxford University Press, 2002, ISBN 0-19-851786-6)

Additional:

1. *The Solid State*, H.M. Rosenberg, (Oxford University Press, 2003, ISBN 0-19-851870-6)
2. *Band Theory and Electronic Properties of Solids*, J. Singleton, (Oxford University Press, 2003, ISBN 0-19-850644-9)
3. *Solid State Physics*, N.W. Ashcroft and N.D. Mermin, (International Thomson Publishing, ISBN 0030839939)

Very Advanced:

1. *Quantum Theory of Many-Particle Systems*, J.D. Walecka and A.L. Fetter, (McGraw Hill College, 1971, ISBN 0070206538, also Dover Pubns, 2003, ISBN 0486428273)

Transport:

1. *Electronic Transport in Mesoscopic Systems*, S. Datta, (Cambridge University Press, 1995, ISBN 0521416043)

Density Functional Theory:

1. *Density-Functional Theory of Atoms and Molecules*, R.G. Parr and W. Yang, (Oxford University Press, 1994, ISBN 0195092767)

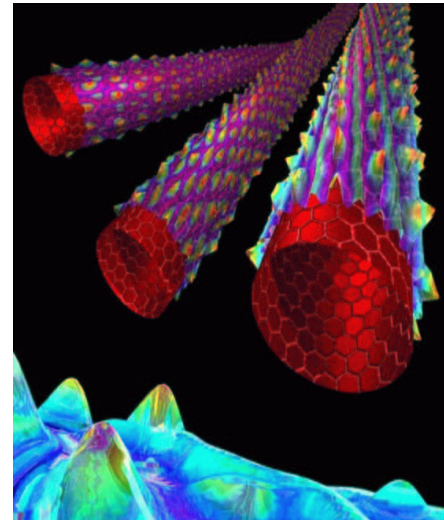
Other resources:

1. Research and Review Articles (distributed in class)

Why doing Condensed Matter Theory?

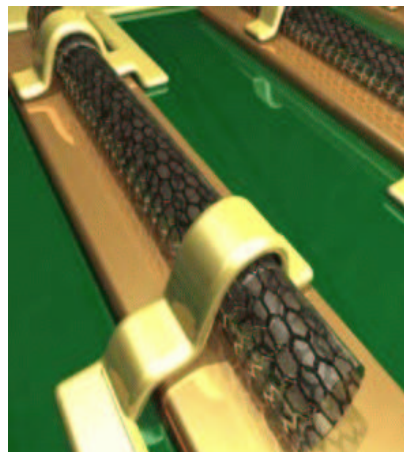
The main idea is to solve REAL problems!!!

- Is it a metal or a semiconductor?
- Why it is so tough?
- How do electrons propagate?



and to make predictions!!!

- Can it be magnetic?
- Can I build a transistor with it?
- How can I use it?
- What if



What Hamiltonian?

You have to decide the minimal Hamiltonian for the specific problem

1. Free-electrons

$$\mathcal{H} = -\frac{\hbar^2 \nabla^2}{2m}$$

2. Electrons in periodic potential

$$\mathcal{H} = -\frac{\hbar^2 \nabla^2}{2m} + V(\vec{r})$$

3. Hubbard Hamiltonian

$$\mathcal{H} = \epsilon_0 \sum_i n_i - t \sum_{i,j,\sigma} \left[c_{i\sigma}^\dagger c_{j\sigma} + c_{j\sigma}^\dagger c_{i\sigma} \right] + U \sum_i n_{i\uparrow} n_{i\downarrow}$$

4. Heisenberg Hamiltonian

$$\mathcal{H} = \sum_{i,j} J_{ij} \vec{S}_i \cdot \vec{S}_j$$

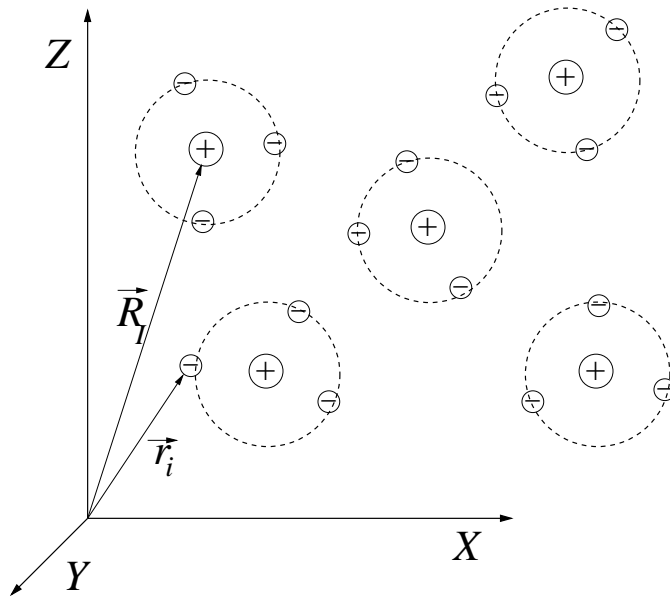
5. Ising Hamiltonian

$$\mathcal{H} = \sum_{i,j} J_{ij} S_{zi} S_{zj}$$

6. Hamiltonian

$$\mathcal{H} = \dots$$

Consider a system of N electrons and M nuclei.



Quantum Mechanics has given us a method to write the general Hamiltonian for such a system. Simply consider the classical Hamiltonian (*non-relativistic*):

$$H_{\text{classical}} = \sum_J \frac{P_J^2}{2M_J} + \frac{1}{2} \sum_{I \neq J}^M \frac{e^2 Z_I Z_J}{R_{IJ}} + \sum_j \frac{p_j^2}{2m_j} + \frac{1}{2} \sum_{i \neq j}^N \frac{e^2}{r_{ij}} - \frac{1}{2} \sum_{i,I}^{N,M} \frac{e^2 Z_I}{R_{iI}}$$

where ($R_{IJ} = |\vec{R}_I - \vec{R}_J|$ $R_{iI} = |\vec{r}_i - \vec{R}_I|$ $r_{ij} = |\vec{r}_i - \vec{r}_j|$)

R_J, P_J position and momentum nucleus J

r_j, p_j position and momentum electron j

Then Quantum Mechanics gives the prescription:

$$\begin{array}{ccc}
 \vec{R} & & \vec{R} \\
 \vec{P}_i & \longrightarrow & -i\hbar \frac{\partial}{\partial R_i} \\
 \vec{r} & & \vec{r} \\
 \vec{p}_i & & -i\hbar \frac{\partial}{\partial r_i}
 \end{array}$$

Therefore the **GENERAL** Hamiltonian for a quantum mechanical system of N electrons and M nuclei is:

$$H(\vec{r}, \vec{R}) = T_N(\vec{R}) + V_{NN}(\vec{R}) + T_e(\vec{r}) + V_{ee}(\vec{r}) + V_{eN}(\vec{r}, \vec{R})$$

$$T_N(\vec{R}) = - \sum_J^M \frac{\hbar^2 \nabla_J^2}{2M_J}$$

Nuclei kinetic Energy

$$V_{NN}(\vec{R}) = \frac{1}{2} \sum_{I \neq J}^M \frac{e^2 Z_I Z_J}{R_{IJ}}$$

Nucleus-Nucleus Interaction

$$T_e(\vec{r}) = - \sum_j^N \frac{\hbar^2 \nabla_j^2}{2m_j}$$

Electrons kinetic Energy

$$V_{ee}(\vec{r}) = \frac{1}{2} \sum_{i \neq j}^N \frac{e^2}{r_{ij}}$$

Electron-Electron Interaction

$$V_{eN}(\vec{r}, \vec{R}) = -\frac{1}{2} \sum_{i,I}^{N,M} \frac{e^2 Z_I}{R_{iI}}$$

Electron-Nucleus Interaction

You can divide H as follows

$$H(\vec{r}, \vec{R}) = H_e(\vec{r}) + H_N(\vec{R}) + V_{eN}(\vec{r}, \vec{R})$$

where we have introduced the nuclei

$$H_N(\vec{R}) = T_N(\vec{R}) + V_{NN}(\vec{R})$$

the electrons

$$H_e(\vec{r}) = T_e(\vec{r}) + V_{ee}(\vec{r})$$

and the interaction Hamiltonian

$$V_{eN}(\vec{r}, \vec{R})$$

Consider the size of the problem. Suppose we are describing $\sim 7\text{gr}$ of Li. This means we have to consider:

$$M \text{ nuclei} \sim 6 \times 10^{23}$$

$$N \text{ electrons} \sim 1.8 \times 10^{24}$$

So in order to describe properly 5gr of Li I have an equation with

$$\underline{3 \times [6 \times 10^{23} + 1.8 \times 10^{24}] = 7.2 \times 10^{24} \text{ variables}}$$

Then the wave function is

$$\Psi(\{\vec{R}\}, \{\vec{r}\}) \equiv \Psi(\vec{R}, \vec{r})$$

and the Schrödinger equation

$$H(\vec{r}, \vec{R})\Psi(\vec{R}, \vec{r}) = E\Psi(\vec{R}, \vec{r})$$

Obviously we must somehow reduce the problem.

*Note that the dynamics of the electrons depends on that of the nuclei and viceversa, **NOT** only on their relative positions*

Born-Oppenheimer Approximation

The nuclei (usually) move at a much slower speed than the electrons

- Nuclei: Sound Speed $v_S \sim 10^2 - 10^3$ m/s
- Electrons: Fermi Velocity $v_F \sim 10^5 - 10^6$ m/s

Electrons can always “adjust” their dynamics to that of the nuclei

$$\Psi(\{\vec{R}\}, \{\vec{r}\}) = \phi(\{\vec{R}\})\psi(\{\vec{r}\}; \{\vec{R}\}) \equiv \phi(\vec{R})\psi(\vec{r}; \vec{R})$$

Electrons know about the nuclei only through their positions. So one can write:

$$H(\vec{R}, \vec{r})\Psi(\vec{R}, \vec{r}) = \phi(\vec{R})H_e(\vec{r})\psi(\vec{r}) + \psi(\vec{r})H_N(\vec{R})\phi(\vec{R}) + V_{eN}(\vec{r}, \vec{R})\phi(\vec{R})\psi(\vec{r})$$

This is still not separable, however since $v_S \ll v_F$ we can write:

$$\left[T_e(\vec{r}) + V_{ee}(\vec{r}) + V_{eN}(\vec{r}, \vec{R}) \right] \psi(\vec{r}) = E(\vec{R})\psi(\vec{r})$$

$$\left[T_N(\vec{R}) + V_{NN}(\vec{R}) + E(\vec{R}) \right] \phi(\vec{R}) = E_{tot} \phi(\vec{R})$$

One solves for electrons first (for each nuclei positions) and then for nuclei.

Born-Oppenheimer Approximation in more details

$$H_e \psi \phi + H_N \psi \phi + V_{eN} \psi \phi = E_{tot} \psi \phi$$

$$[(H_e + V_{eN}) \psi] \phi + \psi H_N \phi = E_{tot} \psi \phi$$

then the Born-Oppenheimer approximation says:

$$(H_e + V_{eN}) \psi = E(\vec{R}) \psi$$

from which we obtain

$$\left[H_N + E(\vec{R}) \right] \phi = E_{tot} \phi$$

Is the problem easier?

- YES, electrons and the nuclei can be treated almost independently
- Not all the nuclei positions are relevant
- Usually nuclei dynamics can be replaced by classical equations
- If I know where the nuclei are I can forget the last equation

Is the problem easy?

NO, I still have 10^{24} electrons

$$\left[T_e(\vec{r}) + V_{ee}(\vec{r}) + V_{eN}(\vec{r}, \vec{R}) \right] \psi(\vec{r}) = E(\vec{R})\psi(\vec{r})$$

In actual fact the REAL problem is the Coulomb term. Suppose it can be neglected then we have

$$\left[T_e(\vec{r}) + V_{eN}(\vec{r}, \vec{R}) \right] \psi(\vec{r}) = E(\vec{R})\psi(\vec{r})$$

The Hamiltonian is then separable and we have a set (1.8×10^{24}) of identical equations:

$$-\frac{\hbar^2 \nabla^2}{2m} \psi(\vec{r}) - \frac{1}{2} \sum_I^M \frac{e^2 Z_I}{r_I} \psi(\vec{r}) = E \psi(\vec{r})$$

With the Coulomb interaction the dynamics of the i -th electron depends on that of the other ($1.8 \times 10^{24} - 1$) electrons

In what follows:

1. We will always use the Born-Oppenheimer approximation
2. In most of the cases we will neglect electron-electron interaction
3. At the end we will put electron-electron interaction back

In conclusion our present problem is to solve an equation of the form

$$-\frac{\hbar^2 \nabla^2}{2m} \psi(\vec{r}) - \frac{1}{2} \sum_I^M \frac{e^2 Z_I}{r_I} \psi(\vec{r}) = E \psi(\vec{r})$$